

# DENSITY SCREEN SCALE

THE DENSITY SCALE BY GOLDANALYTIX

**Instruction manual**

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## 1. About Goldanalytix / Contact

Goldanalytix, founded in the year 2010, is the leading supplier of precious metal testing methods in Germany. Our team works for you to develop safe and reliable testing methods for all kinds of precious metals. The product development and production is carried out completely in Regensburg, Germany. The synergy of analytical know-how and product development allows us to always be up-to-date. Thanks to our steady improvements, we can guarantee highest quality standards.

Do you need data about our products, support on how to use the device or technical support? No problem. You can reach us on many ways:

In the Internet: [www.gold-analytix.com](http://www.gold-analytix.com)

Via e-mail: [gold-analytix@marawe.eu](mailto:gold-analytix@marawe.eu)

Via telephone: +49 941 29020439

We are looking forward to hearing from you!

## 2. Introduction

Congratulations on your purchase of the Goldanalytix DensityScreenScale, the tried and tested device for establishing the density of solid matters.

Please read this instruction manual thoroughly before first using the DensityScreenScale. Measuring the density as the only method is not enough for an absolutely certain result about the authenticity of gold, silver, platinum, palladium and other precious metals. Please always test with at least one further method to exclude precious metal fakes. You can find the most up-to-date version of these instructions at [gold-analytix.com/DensityScreenScale](http://gold-analytix.com/DensityScreenScale) by clicking on "information", which helps you to always be up-to-date after the purchase.

The density test gives you hints about eventual material falsification in cases in which the precious metal has been provided with material of differing density or in case of inferior alloys (e.g. 14 carats gold instead of 21 carats). In these cases, the DensityScreenScale by Goldanalytix will give you precise results when used properly.

However, there are also metals and alloys whose densities resemble those of gold or silver significantly. Gold has a density of  $19.30 \text{ g/cm}^3$ , which is very close to tungsten's density with its  $19.32 \text{ g/cm}^3$ . Silver has a density of  $10.49 \text{ g/cm}^3$ . An alloy of the two metals tin and lead in correct proportions has one of about  $10.50 \text{ g/cm}^3$ . Please keep in mind that these miniscule differences cannot be detected by a density scale, which is why one has to turn to an additional test for e.g. a fake gold bar with tungsten. A good example for detecting this tungsten inclusion would be the ultrasound test which is included in our Goldanalytix bar testing set GAX1000.

### 3. Scope of supply

Your Goldanalytix DensityScreenScale set contains the following components:



#### Precision scales

#### Density measurement assembly, consisting of:

- Water container (Plexiglas)
- Base object (Plexiglas)
- Measurement tray holder (Plexiglas)
- Measurement tray (stainless steel)
- Lid (Plexiglas)

#### Bag of small components, consisting of:

- 4 fixing rods
- 4 oval-head screws
- 1 aluminum cone
- 1 fixing bolt

#### Flash drive with calculation tool




#### Pair of tweezers


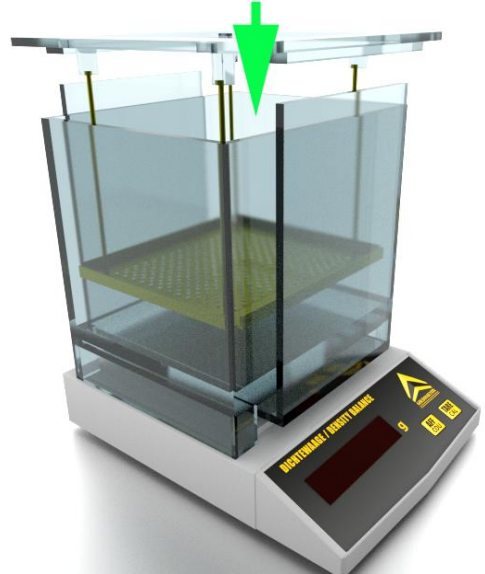
#### Power supply

#### Instruction manual

Should this device or its components be damaged or should some parts be missing, please contact us immediately (contact information, see p. 2).

## 4. Assembly of the DensityScreenScale

Step	Description	
1	Put the scales on a firm and stable surface.	
2	Put the aluminum cone with light pressure on the scales' cone with the screw thread.	
3	Fix the base object made of Plexiglas with the included bolt (in the bag) on the aluminium cone as shown in the picture. Please guarantee that the vertical surfaces are arranged in the way shown in the picture. Pull out the bolt cautiously with a screwdriver. Please be very cautious here and use as little pressure as possible to prevent the scales from being damaged.	

<p>4</p>	<p>Put the empty water container on the scales as shown in the picture. Please guarantee that the container is positioned to fit in the ducts on the scales.</p>	
<p>5</p>	<p>After that, put the already assembled measurement tray holder on the base object. Please guarantee that the ducts in the tray holder are positioned precisely into the lateral walls to guarantee a firm grip of the tray holder. The scales are now ready for use!</p>	

## 5. Preparing the measurement

We recommend you to respect these instructions in order to avoid inaccuracies while establishing the density.

### **Operating time of the scales**

Turn on the scales 5 to 10 minutes before the measurement to guarantee highest precision. Like this, the scales give you the best results because a temperature adjustment of the fine mechanics is needed. Also guarantee a firm place without risk of shaking for the scales. The results of every precision scales are disturbed significantly by shaking, air drafts and fluctuations in temperature.

### **Ambient air temperature**

The density scales have been proven reliable in most surroundings, where a room temperature of 20 to 25 °C is ideal. While the device has been shown to work at temperatures above 35 °C and below 15 °C, one normally avoids such temperatures. The important thing is that the temperature during the operating time stays constant.

### **Temperature of the bath**

It is ideal to use water with a temperature of about 25 °C for the water bath. Do not fill in water that is either very cold or very warm, as this influences the precision of the density measurement. In case of water at room temperature, you can simply use the simplified formula for calculating the density.

### **Sample**

The sample has to be dry and clean and should not have a layer soluble in water. Layers of grease and oil have to be removed before the test.

### **Insertion depth**

Please guarantee that the liquid level between the corresponding establishments of the air value (L) and the water value (W) does not change significantly. The rise of the water level caused by the sample is only important for objects with a big volume and has to be kept in mind accordingly.

### **Bubbles**

In case of liquids that cover feebly (e.g. normal water) it may happen that air bubbles settle down on the support or the sample. A bubble of a diameter of 2 mm, there can already be an inaccuracy of 4 mg. In order to avoid air bubbles, please follow the following steps:


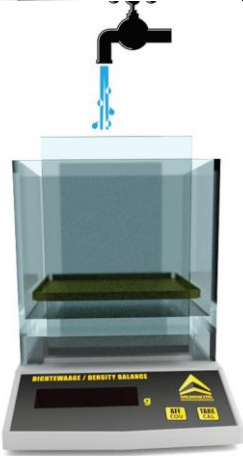

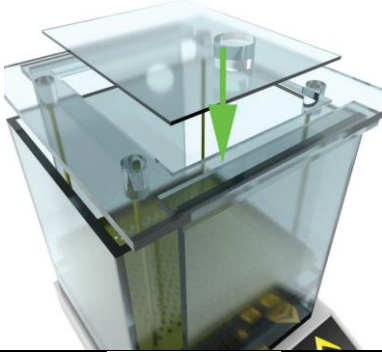



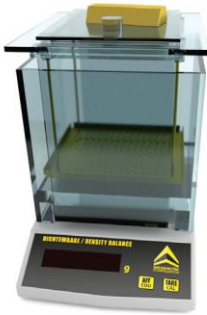

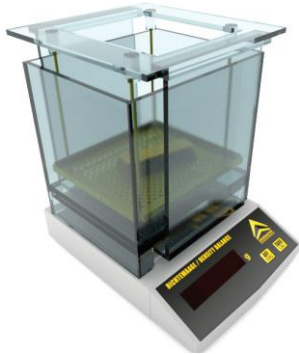
- Degrease the sample and the measurement tray
- To remove the air bubbles, shake the whole assembly with caution after the first immersion
- Remove more stubborn bubbles with a small brush
- Clean the measurement tray regularly

### **Porosity of solid matters**

When dipping solid matters into a liquid, not the entire air is normally replaced in the (micro-) pores. This leads to inaccuracies and therefore to differing densities in case of porous solid matters.

## 6. Carrying out the measurement

Step	Description	
1	Before the measurement: Turn on the scales and take out the measurement tray (if in the device).	
2	Fill the container with pure, clean water. The tank's complete filling volume is 1.5 litres, so please fill in about 1.2 to 1.3 litres. In case of voluminous objects with very high displacement, please fill in bit less.	
3	Dip the tray holder slowly into the water and put it on the tray plate. In case of bigger air bubbles, please remove them carefully with a brush. Please absolutely guarantee that no water drops stick to a part pushing on the measurement cell! All these components have to stay dry.	
4	Put on the lid.	
5	Tare the scales.	

<p><b>6</b></p>	<p>Lay the sample, e.g. a gold bar, on the lid to establish the air value (<b>L</b>). Write this value down.</p> <p>In our example, it is 1000.13 g.</p>	
<p><b>7</b></p>	<p>Take the sample and the lid off the scales.</p>	
<p><b>8</b></p>	<p>Tare the scales.</p>	
<p><b>9</b></p>	<p>Lay the object on the tray using the included tweezers and establish the water value (<b>W</b>). It is important that the whole object is under water. Please wait until the values are stabilized. In our example, this was 948.31 g. This means that the <i>difference</i> between <b>L</b> and <b>W</b> is 51.82 g. The air value <b>L</b> divided by <i>this value</i> equals the <i>density</i>, for example, that means 19.32 g/cm<sup>3</sup>.</p>	
<p><b>10</b></p>	<p>Calculate the density value according to the simple, opposite formula with a calculator.</p>	<p><math>\rho = L / (L - W)</math>          Unser Beispiel:  <math>\rho = 1000,13 / (1000,13 - 948,31)</math>  <math>= 19,32</math>          --&gt; exakt die Dichte von Reingold</p>
<p><b>11</b></p>	<p>For every other measurement, please restart at step 4.</p>	

Alternatively to your own calculation (step 10), you can use the calculation tool “Goldanalytix density calculation” saved on the provided flash drive. This tool is designed especially for the DensityScreenScale and allows you a quick calculation of the density value including short instructions for the density measurement.

All of the density tables are also included in the software. Additionally, you can easily save the values with one click in document.

## 7. Goldanalytix Density Calculation Tool

The provided flash drive includes a calculation tool (Microsoft Excel). With the help of the Goldanalytix Density Calculation Tool you can easily determine the density of your testing objects. The head of the calculation tool contains a short instruction manual which gives you the most important steps for determining the density of solid materials.

The second part provides the calculation tool. Please fill in the entry fields **“Weight in the air”** (exemplarily 31.10 g) and **“Weight in water”** (exemplarily 28.14 g) according to steps 3 and 7 of the short instruction manual. The calculated density will be given in the result field **“Density”** (exemplarily 10.51 g/cm<sup>3</sup>). In case you need some more information you will find this instruction manual (latest version) as pdf by clicking on the hyperlink **“Detailed Instruction Manual”** (internet connection required).

**Goldanalytix Density Calculation Tool**

GOLDANALYTIX

**Short Instruction Manual**

1. Tare the DensityScreenScale.
2. Put the object on top of the lid.
3. Write down the result in the red box "Weight in the air".
4. Take the object off the lid.
5. Tare the DensityScreenScale.
6. Put the object on the measurement tray in the water.
7. Write down the result in the blue box "Weight in water".

**Weight in the air:** 31.10 g

**Weight in water:** 28.14 g

**Manual:** On [Detailed Instruction Manual as PDF HERE](#)

8. The density is displayed in the green box.
9. Read the desired value out of one of the tables below.
10. Compare the desired value with your density result.

**Density:** 10.51 g/cm<sup>3</sup>

*Continuation of the tool's head:*

Below the head of the calculation tool you will find extensive tables giving density values of several alloys (gold, silver, platinum, and palladium) as well as some pure metals. You can assign the calculated density value of your testing object to the values of the tables and compare it to the presumed alloy of your testing object.

Finally, the bottom of the calculation tool contains a table (see below) in which you can enter all of your tested objects, the determined density, and some other information.

Value Table	Name of the tested object	Type of Material	Desired value	Density (g/cm <sup>3</sup> )	Comment

## 8. Interpretation and evaluation of the results

The density is defined as the quotient of the mass and the volume of a material.

$$\rho = \frac{m}{V}$$

The density, according to the international system of units, has the SI-unit kg/m<sup>3</sup> or g/cm<sup>3</sup> (SI: International System of Units, French: *Système international d'unités*) and the symbol ρ (Rho).

### Theory of the density establishment of solid matters

By using a liquid of known density ρ<sub>o</sub> (mostly water), one can establish the density of the solid matter dipped into it. The solid matter is first weighed at air (**L**) and then in the water (**W**). The density can be established in the following (simplified) formula.

$$\rho = \frac{L}{L - W}$$

This formula allows you to calculate the density by yourself.

**Example:** For a gold coin (1 ounce) made of 999.9 pure gold, an air value **L** of 31.13 g has been established. Afterwards, the water value **W** has been established at 29.52 g. Now, you subtract 29.52 g (i.e. the water value **W**) from the 31.13 g (air value **L**). This equals 1.61. This value is divided by the air value **L**, i.e. 31.13 divided by 1.61, which gives you the density value of 19.32 g/cm<sup>3</sup>. This corresponds to the density of pure gold.

Compare the received values with the values you can find in the comparison tables on the next two pages.

Please keep in mind that the value given out by the DensityScreenScale can show deviations from the real density value depending on the mass of the sample. This is fundamentally determined by the principle the measurement is based on. The smaller and denser the sample, the bigger the deviations will be and more water will be displaced by the object. One ounce of silver is therefore easier to examine with a smaller deviation range than e.g. one ounce of pure gold (see picture 1).

This means for example that a differentiation of e.g. sterling silver (10.40 g/cm<sup>3</sup>) and pure silver (10.49 g/cm<sup>3</sup>) is not possible until a certain weight because the theoretical deviations will be small enough. Please keep these limitations of the density measuring principle in mind when evaluating the results.

Please still keep in mind that the densities – especially of 750 to 900 gold alloys – can be achieved by different alloys of metals like tantalum, tungsten, molybdenum and lead and these gold alloys are often imitated by forgers.

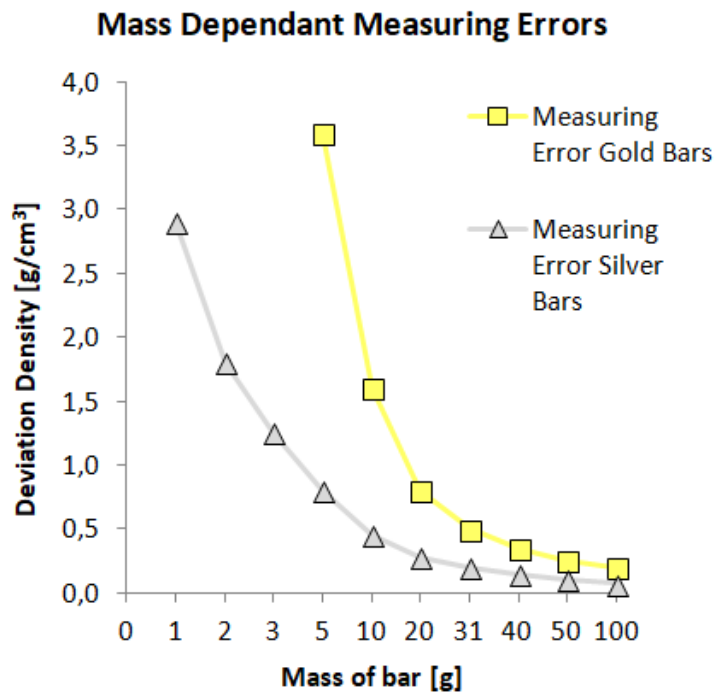


Figure 1. Density scales inaccuracies of gold- and silver bars depending on the mass.

## 9. Density tables of metals and alloys

Alloy	Density [g/cm <sup>3</sup> ]	Metal	Density [g/cm <sup>3</sup> ]
<b>Yellow gold alloys (standard alloys)</b>		Iridium	<b>22.6</b>
		Osmium	<b>22.6</b>
999	<b>19.3</b>	Platinum	<b>21.5</b>
986	<b>19.0</b>	Gold	<b>19.3</b>
916	<b>17.8</b>	Tungsten	<b>19.3</b>
850	<b>16.8</b>	Uranium	<b>19.1</b>
800	<b>16.2</b>	Mercury	<b>14.3</b>
750	<b>15.4</b>	Palladium	<b>12.0</b>
416	<b>11.5</b>	Lead	<b>11.3</b>
375	<b>11.2</b>	Silver	<b>10.5</b>
333	<b>10.9</b>	Molybdenum	<b>10.2</b>
<b>Silver alloys (standard alloys)</b>		Bismuth	<b>9.8</b>
		Cobalt	<b>8.9</b>
999	<b>10.5</b>	Copper	<b>8.9</b>
925	<b>10.4</b>	Nickel	<b>8.9</b>
900	<b>10.3</b>	Cadmium	<b>8.7</b>
835	<b>10.2</b>	Iron	<b>7.9</b>
800	<b>10.1</b>	Manganese	<b>7.5</b>
720	<b>10.0</b>	Indium	<b>7.3</b>
<b>Bullion coins</b>		Chromium	<b>7.2</b>
Maple Leaf	<b>19.3</b>	Zinc	<b>7.1</b>
Philharmonic	<b>19.3</b>	Antimony	<b>6.7</b>
Yuan Panda	<b>19.3</b>	Zirconium	<b>6.5</b>
American Eagle	<b>17.5</b>	Vanadium	<b>6.1</b>
Krugerrand	<b>17.5</b>	Tin	<b>5.8</b>
		Titanium	<b>4.5</b>
		Aluminum	<b>2.7</b>
		Beryllium	<b>1.8</b>
		Magnesium	<b>1.7</b>

Table 1 – Overview of the density values [g/cm<sup>3</sup>] of pure metals and alloys

<b>Gold 333 alloys</b>					
<b>Alloy</b>					<b>Density [g/cm<sup>3</sup>]</b>
<b>Ag in ‰</b>	<b>Cu in ‰</b>	<b>Zn in ‰</b>	<b>Sn in ‰</b>	<b>Ni in ‰</b>	
534	133	0	0	0	<b>11.0</b>
445	222	0	0	0	<b>10.9</b>
333	334	0	0	0	<b>10.9</b>
200	467	0	0	0	<b>10.8</b>
95	572	0	0	0	<b>10.7</b>
114	431	114	0	8	<b>10.8</b>
255	350	47	15	0	<b>11.2</b>
<b>Gold 585 alloys</b>					
<b>Alloy</b>					<b>Density [g/cm<sup>3</sup>]</b>
<b>Ag in ‰</b>	<b>Cu in ‰</b>	<b>Zn in ‰</b>	<b>Cd in ‰</b>	<b>Ni in ‰</b>	
382,5	32,5	0	0	0	<b>13.7</b>
310	35	0	70	0	<b>13.7</b>
280	135	0	0	0	<b>16.6</b>
188	227	0	0	0	<b>15.5</b>
110	184	71	0	50	<b>13.5</b>
90	325	0	0	0	<b>13.4</b>
0	415	0	0	0	<b>13.2</b>
<b>Gold 750 alloys</b>					
<b>Alloy</b>					<b>Density [g/cm<sup>3</sup>]</b>
<b>Ag in ‰</b>	<b>Cu in ‰</b>	<b>Cd in ‰</b>			
250	0	0			<b>15.9</b>
214	36	0			<b>15.8</b>
167	83	0			<b>15.6</b>
125	125	0			<b>15.4</b>
83	167	0			<b>15.2</b>
0	250	0			<b>14.8</b>
167	0	83			<b>15.5</b>

Table 2 – Overview of density values [g/cm<sup>3</sup>] of (non-standard-) gold alloys



## 10. Features of the high precision scales

### Turning on/off (I/O):

On the scales' backside, you can find the main switch for turning the device on or off. The auto-test begins with the display test. When working properly, "8.8.8.8.8" will light up first before the nominal weight ("2000.00 g").

### Warnings

#### *Overload (-----)*

In case of overload, the display will show "-----" (overload). Reduce the load immediately to avoid damages.

#### *Error 1 and Error 2*

Error 1 hints to a swinging load, Error 2 hints to an unstable load without swinging. Please guarantee that the load sits stable on the scales.

#### *Reset (TARE / CAL)*

With TARE, you can set the scales back to 0. This step is essential for e.g. dipping the sample into water and for cases in which the lid is put on the object. Additionally, you can reset the empty weight of a recipient put on top with the CAL-function.

#### *Calibration (CAL)*

By pressing the TARE/CAL-button, you will get to the calibration menu. You will first see --CAL--, as soon as a value lights up, a weight has to be put on the scales. After a short amount of time, the display will be stabilized and you can take off the calibration weights, which concludes the calibration. Please calibrate the scales always after assembling the density scales. You can also use the scales with the included scale tray for normal weight examination.

#### *Counting mode (ZAE)*

In the counting mode, you can find out how often your weight to be measured can be divided in relation to a sample mass. Please put the sample mass on the scales. Change to the corresponding counting mode with the [ZAE]-button in the weighing zone. You will see --COU-- and then the number 10 blinking. With the [TARE/CAL]-button, you can do the corresponding classification (from 10 to 500 in 10, 20, 50, 100, 250, 500 steps). Put the reference weight / number of pieces on the weighing zone. Confirm the configuration with the [ZAE]-button to be able to weigh in the counting mode.

**IMPORTANT:** Please keep in mind that due to the devices own weight, the maximum measurable load for density measurements is reduced accordingly. This is why the density of a 2 kg bar cannot be established.

## 11. Specifications of the high precision scales

Specifications	Description
Net weight (without top part)	about 1.7 kg
Measurement range	2000 g/0.01 g
Additional functions	Auto-calibration Tare-function Counting Error display
Ambient conditions	+5 °C to + 35 °C when functioning +10 °C to + 50 °C when stored

## 12. Environmental and Disposal Instructions



Used electronic devices are not allowed to be deposited in the household waste according to European regulations [1], but have to be disposed separately. The symbol on the dustbin indicates on the necessity of the separation from the household waste. Please help to protect the environment. Please assure that in case of not using the device anymore to give it to the corresponding garbage pick-up.

Please inform yourself about the local waste calendar and your city or municipal administration, respectively, about the opportunities of returning old equipment.

If you need more information please contact us at [gold-analytix@marawe.eu](mailto:gold-analytix@marawe.eu). WEEE- number: DE70793505

[1] Regulation 2002/95/EG of the European Parliament and Council for electronic old equipment

## 13. More non-destructive testing devices by Goldanalytix



### GoldScreenPen

The GoldScreenPen is one of the most versatile electronic precious metal testers. The world's smallest probe tip enables the user to measure coins, ingots and jewellery (even through films and blisters). The measured conductivity value, which is detected up to a depth of 0.5 mm, is displayed on the digital screen.

[www.gold-analytix.com/goldscreenpen-electronic-gold-tester](http://www.gold-analytix.com/goldscreenpen-electronic-gold-tester)

### GoldScreenCard

The GoldScreenCard tests coins and ingots by utilising the eddy current method. The penetration depth is up to 4 mm depending on the metal species. The database, included in the purchased software, features values of most common coins and can be extended by the user with own values – the perfect solution for numismatists.

[www.gold-analytix.com/goldscreen-card-gsc](http://www.gold-analytix.com/goldscreen-card-gsc)



### GoldScreenBox

The GoldScreenBox measures the conductivity of coins and ingots by using the eddy current method with penetration depths up to 650  $\mu\text{m}$ . You can characterize the authenticity of those precious metal objects even through capsules, films and blisters up to a thickness of 3 mm.

[www.gold-analytix.com/goldscreenbox](http://www.gold-analytix.com/goldscreenbox)



### MagneticScreenScale

You can easily identify many counterfeits with the Magnetic Balance, especially the ones made from tungsten. The strong magnetic field of the Magnetic Balance penetrates the ingot or coin without damaging it. Tungsten is detected up to 3 mm beneath the gold plated surface!

[www.gold-analytix.com/magnetic-scale](http://www.gold-analytix.com/magnetic-scale)





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